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**On the structure of uranium in thin film.** By JERRY DONOHUE, *Department of Chemistry, University of Southern California, Los Angeles 7, California, U.S.A.*

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A thin film of uranium, obtained by evaporation of a weighed quantity of the metal from a tungsten bucket inside a vacuum chamber (pressure  $10^{-4}$  mm. Hg) on to a collodion film covering a steel wire-mesh, was examined by the electron diffraction method by Chatterjee (1958). He presented a photograph of the electron-diffraction pattern, and stated that it was found to correspond with the tetragonal structure ( $\beta$  phase) of uranium. The estimated and measured ring diameters, Miller indices and visually estimated intensities were given for 12 observed lines in a table. The estimated diameters were apparently calculated by the use of the lattice constants of  $a = 10.52$ ,  $c = 5.57$  Å at room temperature of a uranium-chromium alloy containing 1.4 at.% of Cr reported by Tucker (1951). (The presence of a small amount of Cr in U suppresses the  $\beta$ -U  $\rightarrow$   $\alpha$ -U transition, which occurs at 660 °C. in pure U. The lattice constants of the alloy were subsequently determined more accurately by Thewlis (1952), who found  $a = 10.590$ ,  $c = 5.634$  Å, but these small differences do not affect the discussion which follows.)

Table 1. *Diffraction data*

<i>hkl</i>	(1) $d_c$	(2) $d_o$	(3) $I_o$	(4) $I$	(5) $pF^2$
311	2.856 Å	2.851 Å	<i>vf</i>	> 4	6
410	2.550			70	78
330	2.480	2.503	<i>f</i>	53	17
202	2.461			41	14
212	2.397	2.322	<i>f</i>	48	36
411	2.320			100	48
331	2.265			58	23
222	2.229			20	8
312	2.135	2.125	<i>vs</i>	19	11
620	1.663	1.707	<i>s</i>	> 4	1
611	1.652	1.632	<i>vf</i>	> 4	> 0.2
{ 522	{ 1.600				
{ 621	{ 1.594			25	33
{ 532	{ 1.514				
{ 631	{ 1.510			28	49
413	1.501			45	37
{ 333	{ 1.486			40	39
{ 602	{ 1.484				

(1) For  $a = 10.52$ ,  $c = 5.57$  Å.

(2) As calculated from the measured diameters given by Chatterjee, with  $\lambda = 0.0480$  Å. Indices are those assigned by him.

(3) As tabulated by Chatterjee.

(4) Observed relative intensities (uncorrected for absorption or L.-P. factors) from the powder data of Thewlis. All lines out to (10,5,1) having  $I > 19$  are included.

(5)  $\Sigma pF^2$ .  $p$  the multiplicity, observed  $F$  values (corrected for absorption and L.-P. factors) calculated from the single crystal data of Tucker & Senio. Average of ( $hkl$ ) and ( $khl$ ) used when possible.

Table 1 (cont.)

<i>hkl</i>	(1) $d_c$	(2) $d_o$	(3) $I_o$	(4) $I$	(5) $pF^2$
612	1.469 Å			42	23
720	1.445			37	42
{ 542	{ 1.415				
{ 641	{ 1.411			21	27
114	1.369	1.357 Å	<i>vs</i>	> 4	> 0.1
{ 314	{ 1.285				
{ 820	{ 1.276			24	41
414	1.222			29	33
424	1.198	1.198	<i>s</i>	> 4	0.3
802	1.189			22	15
742, 812	1.182			22	40
434	1.161	1.166	<i>s</i>	> 4	0.3
{ 723	{ 1.140				
{ 911	{ 1.137			20	25
115	1.102	1.105	<i>vf</i>	> 4	2
860, (10,0,0)	1.052	1.053	<i>vf</i>	> 4	2
{ 932	{ 1.030				
{ 663	{ 1.031			24	31*
{ 10,1,1	{ 1.029				
{ 415	{ 1.021				
{ 10,2,1	{ 1.014			22	23
724	1.003			36	50
{ 952	{ 0.959				
{ 961	{ 0.958			25	1*
804	0.956	0.963	<i>f</i>	24	0.4
10.5.0	0.941			20	11
10.5.1	0.928			23	5

\* Not including the contribution of lines (663) and (961).

Intensity data for uranium powder at 720 °C., and for single crystals of the 1.4 Cr alloy at room temperature have been published by Thewlis (1952) and Tucker & Senio (1953), respectively. Although the data of Chatterjee apparently show fair agreement between observed and calculated spacings, the relative intensities, which are just as important for the identification of a phase, were not used by him. In Table 1 are presented the combined relevant diffraction data of Chatterjee, Thewlis, and Tucker & Senio. It is readily seen that the material which gave the diffraction pattern reported by Chatterjee is not  $\beta$ -uranium, nor does it seem to be  $\alpha$ -U,  $\gamma$ -U,  $UO_2$ ,  $U_3O_8$ , UC, or UN.

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### References

- CHATTERJEE, S. N. (1958). *Acta Cryst.* **11**, 679.  
 THEWLIS, J. (1952). *Acta Cryst.* **5**, 790.  
 TUCKER, C. W. JR. (1951). *Acta Cryst.* **4**, 425.  
 TUCKER, C. W. JR. & SENIO, P. (1953). *Acta Cryst.* **6**, 753.